Superconductivity and phase stability of potassium-doped biphenyl†

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Phenyl molecules are proposed as potential high-temperature superconductors due to exhibiting interesting properties. Here, we report the discovery of superconductivity with the critical temperature \((T_c)\) of \(\sim 7.2\) Kelvin in potassium (K)-doped biphenyl \((\text{C}_12\text{H}_{10})\). The dc magnetic susceptibility measurements provide solid evidence for the presence of the Meissner effect in \(K\text{C}_12\text{H}_{10}\). The Raman spectra detected bipolaronic characteristics in this superconducting state, which are proposed to account for the electron pairing. Theoretical simulations provided the information of the crystal structure of \(K\text{C}_12\text{H}_{10}\). Combining XRD data with formation energy, we suggest that the superconducting phase corresponds to \(K\text{C}_12\text{H}_{10}\) or with a small charge fluctuation in a layered structure. The discovery of superconductivity in K-doped biphenyl vastly expands the potential applications in the superconducting field.

1 Introduction

Organic materials have attracted increasing interest because of their many fascinating phenomena such as low dimensionality, strong electron-electron and electron-phonon interactions and the proximity of antiferromagnetism, insulator states and superconductivity. Especially, organic based compounds were suggested as candidates for high temperature or room temperature superconductors since the electrons can interact with much higher excitation energy than the phonon energy in these materials.1,2 Numerous studies have been done over a half century. In 1965, the first carbon-based superconductor \(\text{C}_6\text{A}\) (A represents the alkali metal) was studied, exhibiting a superconducting transition temperature \((T_c)\) of \(0.02–0.55\) K.3 In 2003, the \(T_c\) values of graphite-like superconductors were raised to 6.5 and 11.5 K for \(\text{YbC}_6\) and \(\text{CaC}_6\), respectively.4,5 In 1991, superconductivity was also observed in the complete carbon material fullerene by doping potassium (K), with \(T_c \sim 18\) K.6 Furthermore, the \(T_c\) in cesium-doped fullerene reached to 38 K.7 In 1980, a real organic superconductor consisting of C and H atoms, \((\text{d}-\text{tetramethyltetrastelenafulvalene})\)-hexauorophosphate, named \((\text{TMTSF})_2\text{PF}_6\), was reported with \(T_c\) equaling 0.9 K under 1.2 GPa.8 Then, in 2003, the \(T_c\) was raised to 13.4 K in the BEDT-TTF salt family under 8.2 GPa.9 Recently, an interesting superconductivity has been discovered in polycyclic aromatic hydrocarbons (PAHs) doped by alkali metals.10–15 The PAH materials exhibit the characteristics of multi-\(T_c\) phases and the highest value of \(T_c\) increases with the number of benzene rings, reaching 33 K in K-doped 1,2;8,9-dibenzopentacene.11 In addition, researchers observed that the polyn-(para-phenylene) (PPP) materials, such as K-doped \(p\)-terphenyl16–20 \(p\)-quaterphenyl,21,22 and \(p\)-quinquephenyl,23 which are non-edge-shared aromatic hydrocarbons, also possess rich superconductivities. It is worth mentioning that \(p\)-terphenyl was found to be superconducting with \(T_c\) as high as 123 K,19 which is comparable to the record high \(T_c\) of about 130 K in some cuprate superconductors at ambient pressure.24 This discovery of high \(T_c\) superconductivity in \(p\)-terphenyl seemingly supports the above idea suggested by Little and Ginzburg.

Interestingly, the low-\(T_c\) phase with \(T_c\) in the range of 5–7 K exists in observed PAH superconductors,10–15 though some PAHs exhibit the characteristics of multi-\(T_c\) phases. By exploring the superconductivity of K-doped solid benzene, which is a basic unit of aromatic compounds (the electron-phonon coupling constant \(\lambda = 0.67\) and \(T_c = 6.2\) K), theoretical studies pointed out that the low \(T_c\) phase of 5–7 K is a common feature for all aromatic hydrocarbon superconductors and is easy to obtain due to better stability.25 The subsequent experimental investigations confirmed the observation, for example, 7.2 K in \(p\)-terphenyl17 and \(p\)-quaterphenyl,22 7.3 K in \(p\)-quinquephenyl,23 7.2 K in 2,2’-bipyridine,26 and 3.5/7.2 K in triphenylbismuth.27 The \(\pi\) electron network plays an important role in the superconductivity of PAH and PPP compounds and drives a strong

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coupling of the charge carriers to the lattices. The electron-phonon interactions can account for the superconductivity of this unified superconducting phase.\textsuperscript{25,28} Within the framework of BCS theory, an increase of doping concentration may result in a stronger electron-phonon interaction, which is helpful for obtaining a higher $T_c$.\textsuperscript{29} However, the electron-phonon interaction is not enough to produce the high $T_c$ superconductivity such as 18 K in picene,\textsuperscript{10} 15 K in coronene,\textsuperscript{11} 33 K in 1,2,8,9-dibenzopentacene,\textsuperscript{13} and 43/123 K in $p$-terphenyl.\textsuperscript{18} Other factors may play important roles in the high $T_c$ superconductivity. Key features of organic superconductors are the low dimensionality of the materials and strong electron-electron correlation. The fulleride Cs$_3$C$_60$ was found to be a true Mott insulator because of strong electronic correlation effects,\textsuperscript{31} though it could be turned into a superconductor by pressure.\textsuperscript{7} Superconductivity induced by pressure in Cs$_3$C$_60$ was found to be related to the antiferromagnetic Mott insulator.\textsuperscript{32} K$_3$ picene was also revealed as a strongly correlated electron system,\textsuperscript{33,34} which results in the absence of metallicity.\textsuperscript{35} Theoretical calculations predicted that K-doped PAHs in their ground state exhibit near antiferromagnetic behaviour.\textsuperscript{34,36,37} Especially for K-doped $p$-terphenyl, the bipolaronic characteristic has been identified from Raman scattering measurements.\textsuperscript{37}–\textsuperscript{39} Photoemission spectroscopy conducted on surface K-doped $p$-terphenyl crystals showed a gap persisting up to at least 120 K that shares a similar temperature dependence to the obtained spectra of Bi–Sr–Ca–Cu–O superconductors.\textsuperscript{38} Furthermore, an analogous gap below 50 K was also found in K-doped $p$-terphenyl films fabricated on Au(111).\textsuperscript{39} All these features indicate that the high $T_c$ superconductivity of organic materials with $\pi$-electron networks cannot simply be explained by the electron-phonon mechanism. This indicates that both the electron-phonon interaction and electron-electron correlations work together to enhance the $T_c$ of aromatic hydrocarbons with increasing numbers of benzene rings.\textsuperscript{35}

Due to the appearance of superconductivity and possible high $T_c$, PPP superconductors provide a good system to study the interplay of strong electron-electron and electron-phonon interactions in a low-dimensional system. Comparing with $p$-terphenyl, biphenyl (C$_{12}$H$_{10}$) is the shortest polymer with only two phenyl rings connected by a single C-C bond. It shows a relatively simple monoclinic structure, and it undergoes an incommensurate triclinic transition at about 40 K.\textsuperscript{40} In this work, therefore, we have synthesized K-doped biphenyl (K$_x$C$_{12}$H$_{10}$) and observed the Meissner effect in this material. The dc magnetic measurements support the existence of superconductivity with $T_c \sim 7.2$ K in this material. Combining with the first-principles calculations, we predicted the crystal structure and electronic band feature of this superconducting phase to understand the superconductivity of K-doped biphenyl. Our results report the smallest $p$-oligophenyl superconductor and present the possible microstructure.

## 2 Experimental and computational methods

### 2.1 Experimental details

High quality potassium and crystalline biphenyl were purchased from Sinopharm Chemical Reagent and Alfa Aesar, respectively. In our experiments, the doped biphenyl was synthetized by heating a mixture of small pieces of fresh potassium and biphenyl with a mole ratio of 3:1 at a temperature of 120°-130 °C for 48 hours. All the synthesis procedures were performed in a glove box with moisture and oxygen levels less than 0.1 ppm, and the sample was sealed in a quartz tube under a high vacuum of better than $1 \times 10^{-4}$ Pa in the heating process. The dc magnetization measurements were performed in the temperature range from 1.8 K to 300 K by a commercial Magnetic Property Measurement System (Quantum Design). In the Raman scattering experiments, a 660 nm excitation was applied to obtain the Raman spectra of the pristine and doped biphenyl. The laser power was selected as less than 2 mW before a $\times 20$ objective to avoid possible damage of samples. The X-ray diffraction experiments were performed in the 2-theta range of 5°–60° by an X-ray diffractometer (Empyrean, PANalytical B.V.). The samples were put on specific glasses and then covered with thin films to isolate the air.

### 2.2 Computational details

To study the structural and electronic properties of pristine and K-doped biphenyl, we employed the Vienna ab initio simulation package (VASP)\textsuperscript{41,42} based on the projectors augmented wave method. For the plane-wave basis-set expansion, an energy cutoff of 600 eV was adopted. The Monkhorst–Pack $k$-point grids were generated according to the specified $k$-point separation of 0.02 Å$^{-1}$ and the convergence thresholds were set as $10^{-6}$ eV in energy and $10^{-3}$ eV Å$^{-1}$ in force. The generalized gradient form (GGA) of the exchange–correlation functional (Perdew–Burke–Ernzerh of 96, PBE) was adopted.\textsuperscript{43} Considering the non-local interactions, we added the correction of van der Waals (vdW) in the version of vdW-DF2 in the calculations.\textsuperscript{44} The necessity of the vdW-DF2 functional has been confirmed by our previous studies.\textsuperscript{29,45}

## 3 Results and discussion

### 3.1 Meissner effect

The Meissner effect and zero resistivity are two fundamental features of superconductivity. Actually, metal-doped carbon-based superconductors were mainly detected from magnetic measurement.\textsuperscript{10–15} However, with its small diamagnetic volume and easily damaged crystal structure, the metallic state of K$_x$C$_{12}$H$_{10}$ was hard to realize even below $T_c$. Difficulties in detecting weak superconductivity from resistivity measurements can be overcome by using magnetic susceptibility experiments. Hence, the superconductivity of K$_x$C$_{12}$H$_{10}$ was mainly characterized by magnetization measurements.
Fig. 1 (a) The temperature dependence of dc magnetic susceptibility $\chi$ for K$_x$ biphenyl in the applied magnetic field of 20 Oe with field-cooling (FC) and zero-field cooling (ZFC). Inset is the black synthetic sample of K$_x$ biphenyl. (b) The temperature dependence of $\chi$ for K$_x$ biphenyl measured at various magnetic fields up to 600 Oe in the ZFC run.

Fig. 2 (a) The magnetic field dependence of magnetization for K$_x$ biphenyl at various temperatures in the superconducting state. (b) The magnetization hysteresis loop with scanning magnetic field along two opposite directions up to 800 Oe measured at various temperatures in the superconducting state.

Fig. 1(a) shows the dc magnetic susceptibility $\chi$ of the powder sample K$_x$C$_{12}$H$_{10}$ [inset of Fig. 1(a)] versus temperature measured in the zero-field cooling (ZFC) and field cooling (FC) cycles at 20 Oe. We can see a sudden drop of $\chi$ at the temperature of 7.2 K in the ZFC run, indicating that a superconducting transition occurs in this sample K$_x$C$_{12}$H$_{10}$. The superconducting transition temperature $T_c$ was determined as 7.2 K. Supposing that the density $\rho$ of this sample is about 3 g cm$^{-3}$, we can obtain the shielding fraction $4\pi\chi/\rho = 0.004\%$ from Fig. 1(a). Such a small shielding fraction is due to the existence of impurities and the smaller size of crystallites compared to the London penetration depth in this powder sample. Noticeably, both ZFC and FC $\chi$ show an upturn trend at a lower temperature than $T_c$, which implies a strong paramagnetic signal. The magnetic susceptibility of K$_x$C$_{12}$H$_{10}$ as a function of temperature in the ZFC run under different magnetic fields was also obtained to demonstrate the superconducting transition of K$_x$C$_{12}$H$_{10}$. As shown in Fig. 1(b), the diamagnetism of K$_x$C$_{12}$H$_{10}$ gradually decreases with increasing magnetic field. That is, the superconducting transition gradually disappears with increasing magnetic field since it is suppressed by the applied magnetic field. However, the superconductivity can be observed until the magnetic field is larger than 500 Oe.

The magnetic field dependence of the magnetization for the superconducting K$_x$C$_{12}$H$_{10}$ at various temperatures from 2 K to 6.5 K with low applied fields is summarized in Fig. 2(a). The linear behavior of the magnetic-dependent magnetization signals the Meissner effect in this superconductor. The superconductivity is suppressed as the temperature increases above 7 K. Fig. 2(b) shows the magnetization hysteresis loop with magnetic field along two opposite directions up to 800 Oe at various temperatures from 2 K to 6.5 K in this superconducting state. The symmetric hysteresis loops indicate that this sample exhibits a strong bulk pinning, which is certification of a type II superconductor.

3.2 Raman spectra

To understand this superconducting material from the aspect of the molecular dynamics process, we measured the Raman scattering spectra of the pristine and the doped biphenyl by using a laser wavelength of 660 nm, respectively shown in Fig. 3. According to previous studies, all of the major peaks in the spectra of pristine biphenyl can be assigned and classified as the ring deformation, ring breathing, C–H bending, and C–C
The two peaks near 1000 and 1250 cm\(^{-1}\) are the C–H bending out-of-plane and the C–H bending in-plane, respectively. The peaks around 1600 cm\(^{-1}\) are the breathing and the inter-ring C–C stretching, respectively. The two bipolaronic bands at 1160 and 1200 cm\(^{-1}\) are due to the intra-ring C–C stretching.

It is well-known that polarons and bipolarons are able to form in conducting polymers. Moreover, bipolaronic characterization in alkali-metal doped PPPs synthesized by different methods has been extensively studied in previous works. The almost separated two intra-ring C–C stretching modes at around 1595 cm\(^{-1}\) in the spectra of pristine biphenyl downshift and merge with bipolaronic bands localized at 1586 cm\(^{-1}\). The observation of the 1475 cm\(^{-1}\) mode with no corresponding band in the parent sample can be considered as the fingerprint for the formation of bipolarons. The 1325 cm\(^{-1}\) and 1360 cm\(^{-1}\) bands of K\(_2\)C\(_{12}\)H\(_{10}\) originate from the inter-ring C–C stretching mode at 1265 cm\(^{-1}\) of the pristine sample. The upshifts in wavenumber reflect the decrease of length in the C–C bonds between rings. The 1260 cm\(^{-1}\) mode present in the Raman spectrum of K\(_2\)C\(_{12}\)H\(_{10}\) is due to the C–H bending in-plane, corresponding to the 1200 cm\(^{-1}\) mode in the pristine sample. The two bipolaronic bands at 1160 and 1200 cm\(^{-1}\) are induced by the C–H bending in-plane, corresponding to the 1155 cm\(^{-1}\) mode in the pristine sample. The bands localized at 965 cm\(^{-1}\) are from the C–H bending out-of-plane. These bands also exhibit bipolaronic characteristics. As a result, the Raman characterization of K\(_2\)C\(_{12}\)H\(_{10}\) proves the existence of bipolarons, which possibly accounts for the observed superconductivity in K\(_2\)C\(_{12}\)H\(_{10}\).

### 3.3 Crystal structure and stability

The identification of crystal structure and the investigation of electronic properties are helpful for exploring this superconductor. Therefore, we carried out first-principles calculations to predict the crystal and electronic structures of K-doped biphenyl. First, we performed a test in the case of the pristine sample. As shown in Table 1, we obtained the optimized crystal parameters of \(a = 11.862 \, \text{Å}, \ b = 5.580 \, \text{Å}, \ c = 7.965 \, \text{Å}, \ \beta = 127.6^{\circ},\) and \(V = 414.0 \, \text{Å}^3\), which are in good agreement with the experimental values. The result implies the feasibility of the theoretical study. Then, we considered four possible doping concentrations, \(x = 1, 1.5, 2,\) and \(3\) in K\(_x\)C\(_{12}\)H\(_{10}\). Selecting the structures with the lowest total energy for each doping level, the optimized crystal parameters are summarized in Table 1 and the geometrical configurations are shown in Fig. 4. Detailed atomic coordinates and bond lengths for pristine and K-doped biphenyl stable structures are presented in the ESI† (Tables S1–S5 and Fig. S1–S5). After doping K metal into biphenyl, great changes took place in the lattice constants, and the volume of the unit cell gradually increased with the doping content. As shown in Fig. 4, the biphenyl molecules respectively form the herringbone configuration in K\(_x\)C\(_{12}\)H\(_{10}\), K\(_{1.5}\)C\(_{12}\)H\(_{10}\), and K\(_2\)C\(_{12}\)H\(_{10}\) similar to K-doped PAHs, however, a typical layered structure is observed in K\(_3\)C\(_{12}\)H\(_{10}\). Noticeably, there is a tilted angle between two successive benzene rings in K\(_2\)C\(_{12}\)H\(_{10}\) but this tilted angle disappears when increasing the K content. Viewing perpendicular to the molecular surface, the two ends of the biphenyl molecule are favorable for K atoms.

To examine the possible doping situation corresponding to our experiment, we analyzed the thermodynamic stability of these considered doping cases by calculating the formation energy \(E_f\). In this work, the \(E_f\) for the doping level \(x\) is defined as a function of K chemical potential with the formula:

\[
E_f = \left( E_x - xE_{\text{H}} - CE_{\text{K}} \right) / (x) = \left( E_x - xE_{\text{H}} - 30.12CE_{\text{K}} \right) / (x)
\]
\[ E_f = E_{\text{doped}} - E_{\text{pristine}} - x\mu_{\text{K}}^\text{bulk} - x(\mu_{\text{K}} - \mu_{\text{K}}^\text{bulk}) \]  

where \(E_{\text{doped}}\) and \(E_{\text{pristine}}\) are the total energy of the doped and host crystal, respectively, \(\mu_{\text{K}}^\text{bulk}\) can be obtained from the energy per K atom in the K metal with the bcc structure, \(x\) is the doping concentration, and \(\mu_{\text{K}}\) is the chemical potential of the K species. Here \(\mu_{\text{K}} = \mu_{\text{K}}^\text{bulk}\) means that the element is so rich that the pure element phase can form. \(E_f < 0\) indicates that the doped compound can stably exist. From the calculated formation energy shown in Fig. 5, all of the considered doped phases are able to exist when the chemical potential of K satisfies a certain condition. Comparing several doping levels, however, K\(_2\)C\(_{12}\)H\(_{10}\) is more stable since it has a lower formation energy than the other phases in a wide range of chemical potential because of the \(E_f < 0\) condition. This suggests that the experimentally observed superconducting phase of 7.2 K is possibly K\(_2\)C\(_{12}\)H\(_{10}\).

### 3.4 X-ray diffraction pattern

The X-ray diffraction (XRD) patterns shown in Fig. 6 further follow the superconducting state observed by experiment corresponding to the doping level of \(x \approx 2\). Comparing the XRD spectra of the pristine and doped biphenyl from experiments, the (100) peak at 9° remains in K-doped biphenyl, which indicates that no big change exists in the lattice constant of \(a\) after doping. A new XRD peak marked as (102) appears at about 16° for this experimentally doped sample that is different from the pristine biphenyl, which means that there is an expansion of lattice constants along the \(c\) direction. Based on these constraints, we first checked the XRD patterns of the predicted structures of K\(_{1}\)C\(_{12}\)H\(_{10}\), K\(_{1.5}\)C\(_{12}\)H\(_{10}\), K\(_{2}\)C\(_{12}\)H\(_{10}\), and K\(_{3}\)C\(_{12}\)H\(_{10}\) shown in Fig. 6. It was found that several new XRD peaks emerge in the range of 10°–16° and the peak at 9° shifts towards a higher angle in the two kinds of doping situations of K\(_{1}\)C\(_{12}\)H\(_{10}\) and K\(_{1.5}\)C\(_{12}\)H\(_{10}\). Moreover, the peak at 9° greatly shifts towards a lower angle in K\(_{2}\)C\(_{12}\)H\(_{10}\). Thus, we can exclude the possibility of K\(_{1}\)C\(_{12}\)H\(_{10}\), K\(_{1.5}\)C\(_{12}\)H\(_{10}\), and K\(_{3}\)C\(_{12}\)H\(_{10}\) from the XRD patterns. Especially, K\(_{2}\)C\(_{12}\)H\(_{10}\) exhibits antiferromagnetism instead of superconductivity. Both K\(_{1.5}\)C\(_{12}\)H\(_{10}\) and K\(_{2}\)C\(_{12}\)H\(_{10}\) possess a higher formation energy, as shown in Fig. 5. The structure of K\(_{2}\)C\(_{12}\)H\(_{10}\) produces XRD peaks near to those of the experimentally doped sample. As shown in Fig. 6, K\(_{2}\)C\(_{12}\)H\(_{10}\) with a small fluctuating charge such as (K\(_{2}\)C\(_{12}\)H\(_{10}\))\(^{+}\)(injected electron) exhibits coincident XRD peaks with the experimental sample, especially the peaks at low angles. Moreover, our previous studies demonstrated the stability and feasibility of doped PAHs with small charge. Therefore, we can conclude that the observed superconducting state corresponds to K\(_{2}\)C\(_{12}\)H\(_{10}\) or with a small fluctuated charge. The crystal structure of this superconducting material is shown in Fig. 4(c).

### 3.5 Electronic structures

Fig. 7(a) shows the band structure of K\(_{2}\)C\(_{12}\)H\(_{10}\) along the high-symmetry \(k\)-points \(I(0, 0, 0), B(0.5, 0, 0), A(0.5, 0.5, 0), Y(0, 0.5, 0), Z(0, 0, 0.5), D(0.5, 0, 0.5), E(0.5, 0.5, 0.5),\) and \(C(0, 0.5, 0.5)\) in the Brillouin zone. Fig. 7(b) shows the total density of states (DOS) and projected density of states (PDOS) of K\(_{2}\)C\(_{12}\)H\(_{10}\). (K\(_{2}\)C\(_{12}\)H\(_{10}\))\(^{+}\) and (K\(_{2}\)C\(_{12}\)H\(_{10}\))\(^{-}\). There is an indirect overlap between conduction bands and valence bands at the Fermi level of K\(_{2}\)C\(_{12}\)H\(_{10}\) shown in Fig. 7(a), which means a small amount of electronic states [0.47 states per eV per f.u. shown in Fig. 7(b)] at the Fermi level. The density of states at the Fermi level is less than 4.6 states.
per eV per f.u. of K$_2$C$_6$H$_6$ where the $T_c$ was predicted as 6.2 K.\textsuperscript{25} Thus, based on the electron-phonon coupling mechanism, the predicted superconducting transition temperature $T_c$ of K$_2$C$_{12}$H$_{10}$ is low. However, with a small fluctuated charge such as (K$_2$C$_{12}$H$_{10}$)$_{+0.1}$e, the density of states at the Fermi level will increase to 3.4 states per eV per f.u. as shown in Fig. 7(b). There will be a big value for more additional charge. Additionally, from the PDOS shown in Fig. 7(b), we observe that the electronic states near the Fermi level are mainly contributed by the C 2p state, which means that the C atoms dominate the superconductivity of K-doped biphenyl.

4 Conclusions

In summary, we have synthesized K-doped biphenyl and carried out the magnetic susceptibility and Raman scattering measurements. Based on the Meissner effect, we found that K-doped biphenyl is superconducting with a $T_c$ of 7.2 K. From the Raman spectra, the bipolaronic characteristics could be observed in this superconducting sample, which are proposed to account for the electron pairing. Furthermore, the crystal structure of this superconducting phase was suggested by employing first-principles calculations. Combining XRD data with formation energy, we suggested that the superconducting phase corresponds to K$_2$C$_{12}$H$_{10}$ or with a small charge fluctuation in a layered structure. Electronic states demonstrate the reasonability of the superconductivity of $T_c \approx 7$ K. The discovery of the superconductivity in K-doped biphenyl is an interesting and significant achievement in organic materials research. This discovery vastly expands the potential applications of superconducting materials.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

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